

Electrochemical Pickling of Steel for Industrial Applications: Modelling

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Introduction: The present work is aimed at *simulating* the steel *electrochemical pickling* process by means of experimental investigation in industrial field and the development of a mathematical *model*. The reference system is characterized by a cell in which an electrolyte is present. A group of electrodes (cathodes and anodes) is immersed in the electrolytic bath. A steel strip, to be submitted to the *pickling* treatment, is interposed between the electrodes.

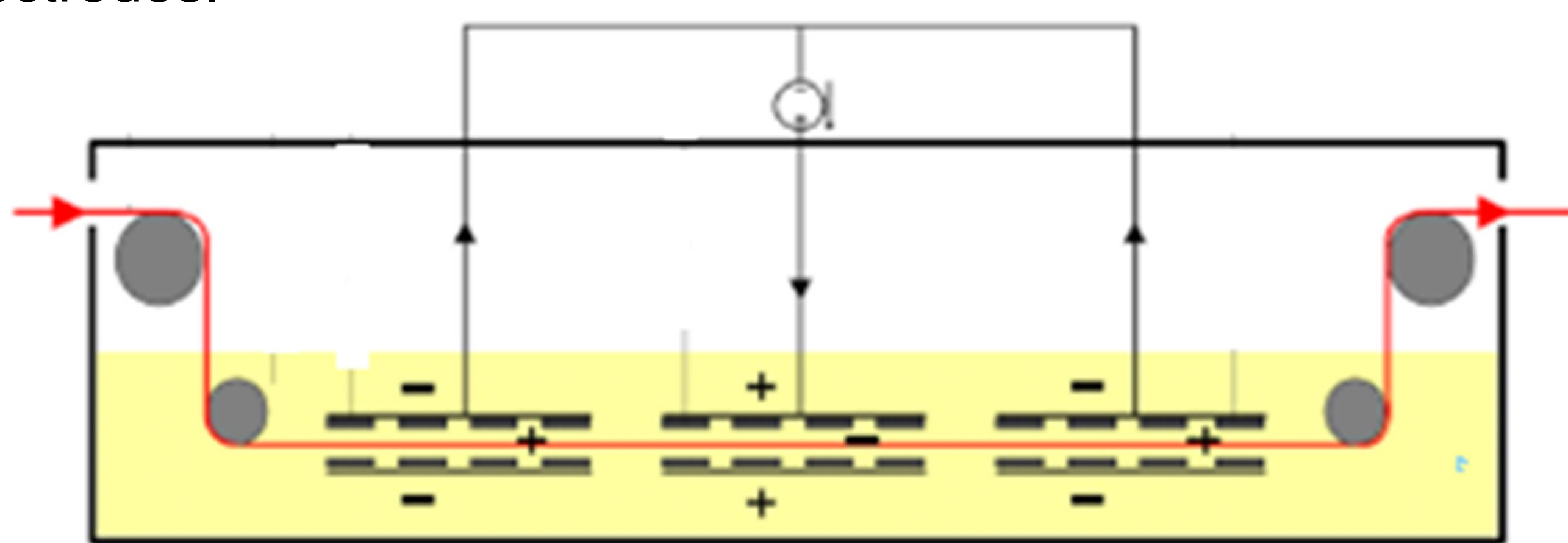


Figure 1. Industrial electrolytic pickling cell

Computational Methods: A reliable, flexible and robust *3D model* has been made for *simulating* the steel strip electrochemical *pickling*. This process is modeled like a *multiphysics system* for the *current control*.

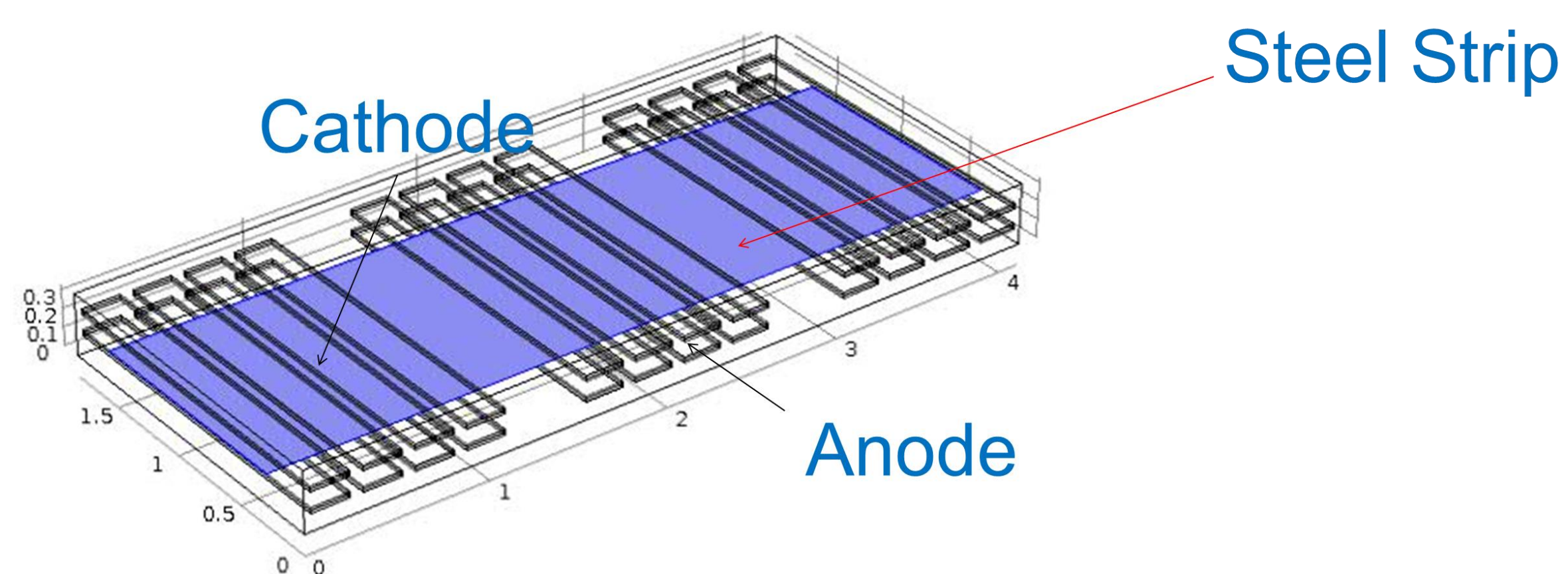


Figure 2. Geometric model of the electrolytic pickling cell

The physics of the problem is described by the following equations system:

$$\begin{cases} \nabla \cdot i_l = Q_l & (1.1) \\ \nabla \cdot i_s = Q_s & (2.1) \end{cases} \quad \begin{cases} i_l = -\sigma \nabla \Phi_l & (1.2) \\ i_s = -\sigma \nabla \Phi_s & (2.2) \end{cases}$$

the equations system is completed by the next set:

- the boundary conditions of the current which develops on both the electrodes (Anode-Cathode) and the strip surfaces :

$$\eta_\Omega = \eta_a = \Phi_s - \Phi_l - E_{eq} \quad (3.1)$$

$$i_k = i_0 \left[e^{\frac{\alpha k_a F \eta}{RT}} - e^{-\frac{\alpha k_c F \eta}{RT}} \right] \quad \forall k = 1, \dots, N \quad (3.2)$$

- the initial condition for the potential in the electrolyte, the electrodes and the steel strip:

$$\Phi_s = V_{s0} \quad (4.1) \quad \Phi_l = V_{l0} \quad (4.2)$$

where the equations:

-1.1 and 2.1 are obtained by the combination of the mass conservation law with the Nernst-Planck's law and Faraday's one in the steady case and applied in the electrolyte, in the electrodes (Anode-Cathode) and in the steel strip

- 1.2 and 2.2 represent the Ohm's law in differential form
- 3.1 represents the secondary current distribution calculated according to the second Kirchoff's law of the equivalent electric circuit
- 3.2 is the empirical Tafel's law for each ionic specie
- 4.1 and 4.2 are the initial potential values in the cell

Results: The obtained *model* allows to estimate, evaluate and verify the operating conditions of a cell dedicated to the industrial electrochemical *pickling* process. According to the logic of management of industrial plant, the total current is the "driving-control" variable. Then it is possible to evaluate:

- the potential distribution in the cell
- the current density distribution in the steel strip
- the geometric effect on the current density distribution
- the voltage-current characteristic curve.

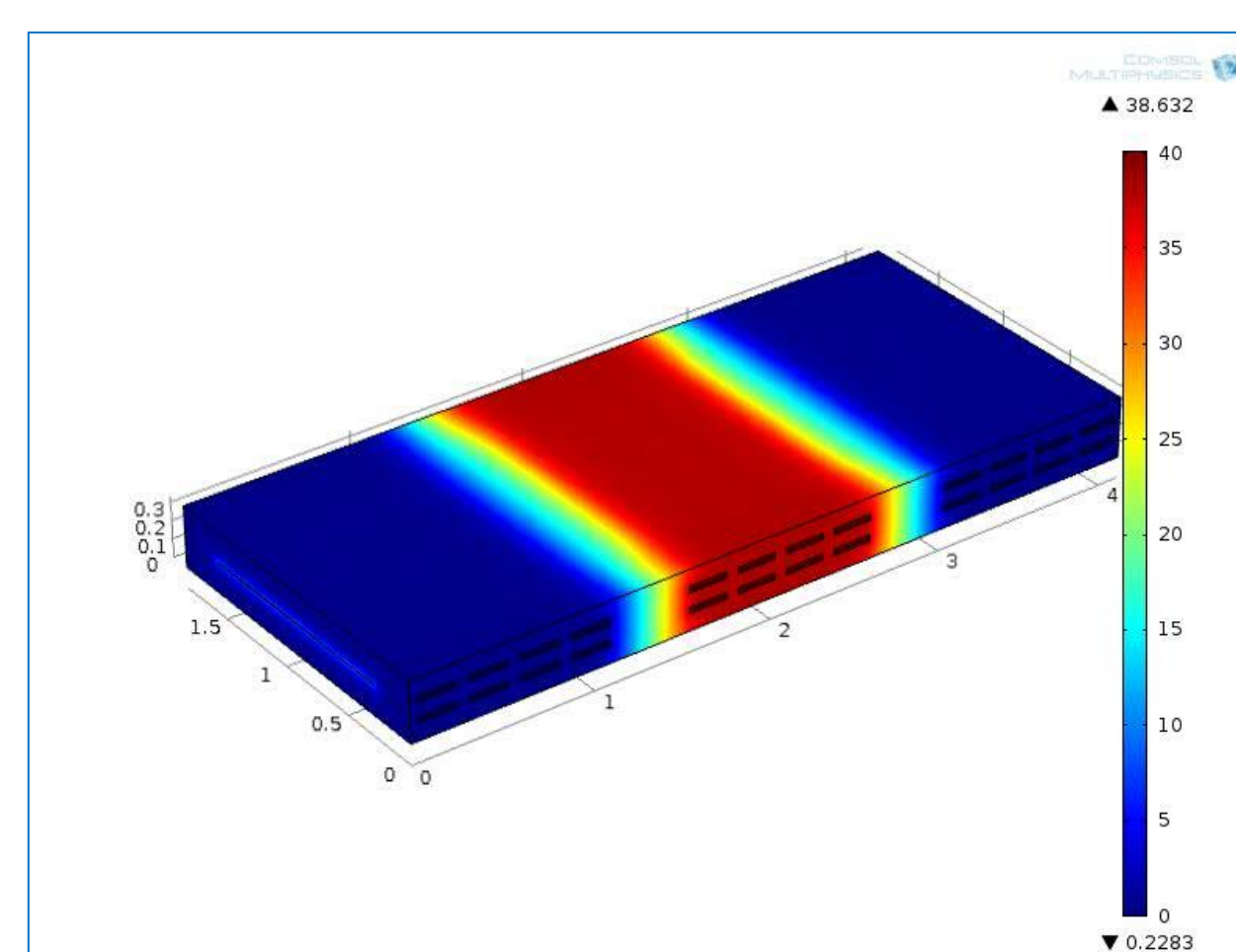


Figure 3. Initial potential distribution in the cell

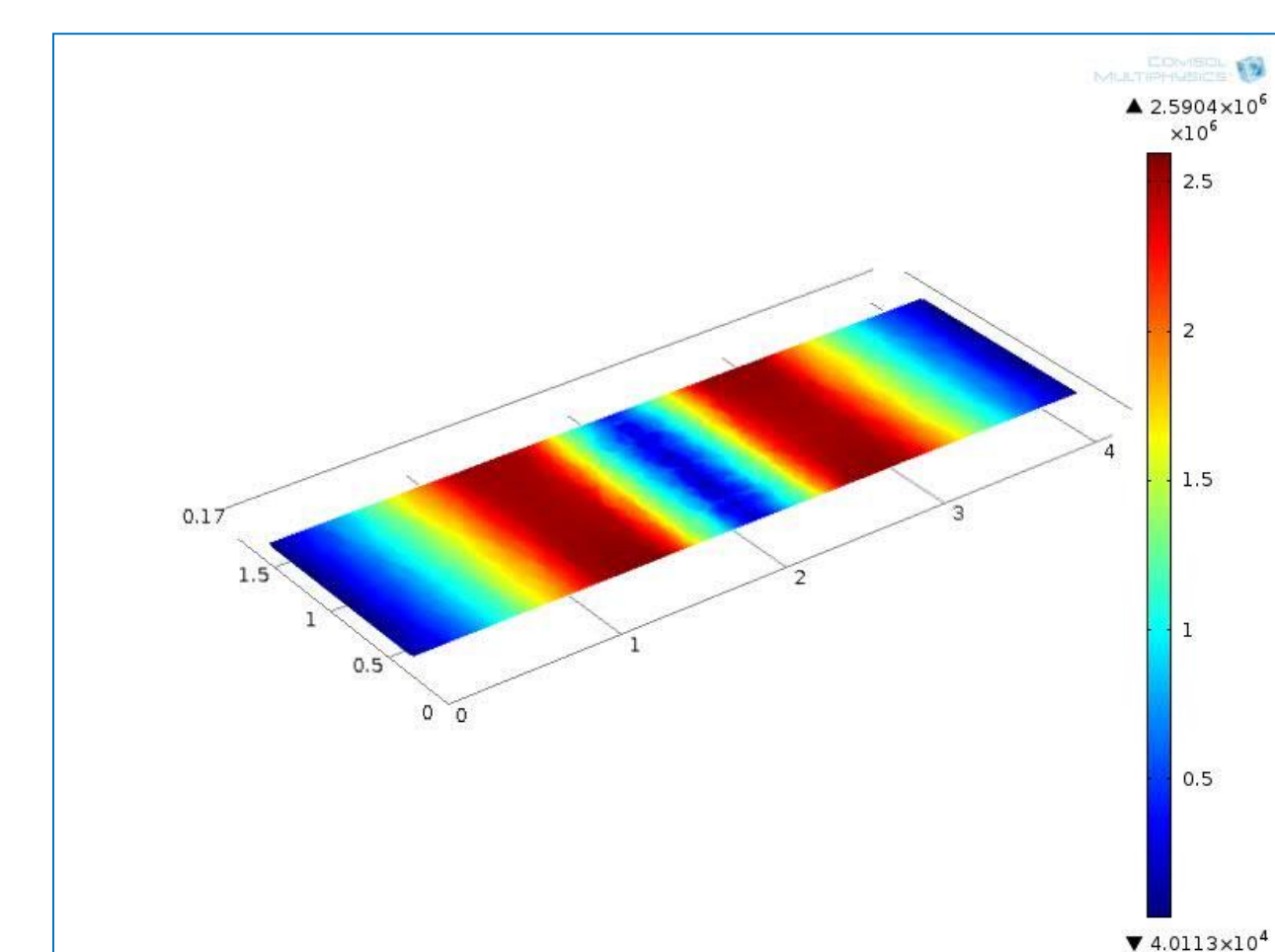


Figure 4. Steel strip current density distribution

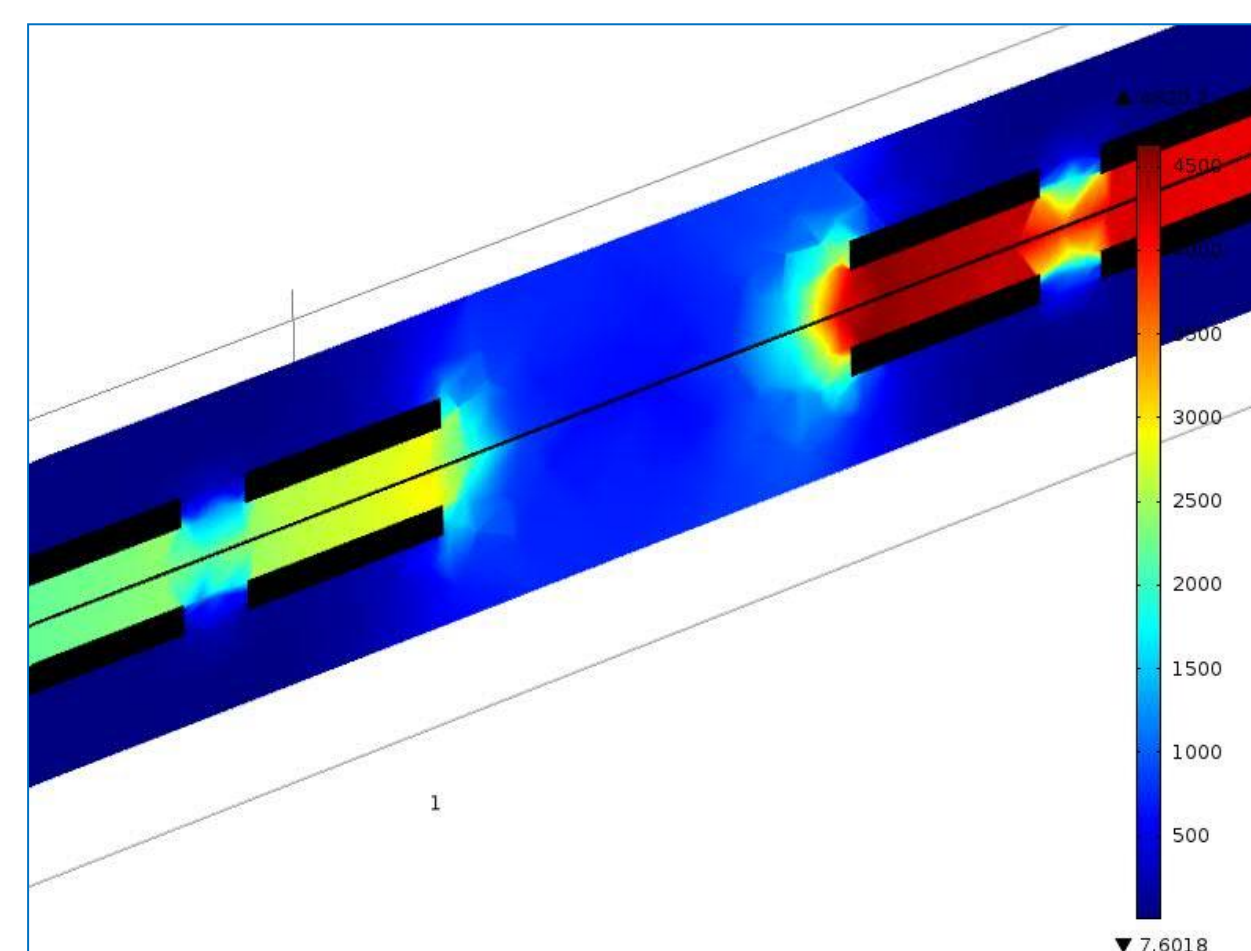


Figure 5. Geometry effect on the current density

I [A]	V [V]
6000	7.33
10000	10.38
14000	13.42

Table 1. Voltage-Current characteristic curve

Conclusions: The system is an excellent tool to use during feasibility study, plants analysis and design, testing and maintenance in the *electrolytic pickling* plant. The *model* results were compared and validated with real process data. Different *pickling* cells/plants will be *simulated* in the future. The system will be upgraded with additional physics (e.g. *electrolysis gas modelling*), as well.

References:

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