Effect of Bed Diffusion and Operating
 Parameters on Char Combustion in the
 Context of Underground Coal Gasification

by

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- Introduction
- Kinetic Determination
- Boat Reactor Experiments
- Modeling of Boat Reactor Experiments
- Conclusions

Kinetic Determination

> Boat Reactor Experiments

Modeling of Boat Reactor Experiments

Conclusions



http://www.dti.gov.uk/energy/sources/renewables/publications/page19148.html

Importance of Combustion

- ➢ It is the essential heat source for endothermic gasification reactions.
- ➢It is the only reaction during early cavity growth.
- ➢So, to study effect of different parameters on combustion reaction becomes a very important part of understanding UCG.

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TGA Experiments for Kinetics



Preliminary experiments conducted with different crucible fillings





Schematic of crucible filling

••• Model Fitting

Conversion vs. time at 600 °C



Random pore model gives the best fitting !!

Kinetic Parameters

BET Results :

BET surface area (S₀) = 179.16 m²/g $\psi = \frac{4\pi L_0 (1 - \varepsilon_0)}{S_0^2} = 3.8485$

From pore size distribution:

$$\mathcal{E}0 = 0.2531$$
;
 $L_0 = 2.754 \times 10^{12}$ cm

 Temperature (°C)
 K $\frac{ksS_0}{(1-\varepsilon_0)}$ (1/sec)

 500
 5x10^{-3}

 550
 1.2x10^{-2}

 600
 4x10^{-2}



k _o	Ea
$0.2 \ge 10^{6} (\text{sec}^{-1})$	116.7 kJ/mol

Kinetic Determination

> Boat Reactor Experiments

Modeling of Boat Reactor Experiments

Conclusions

••• Experimental Setup



••• Experimental Conditions

- > At low temperature & relatively higher flow rates
- Char particle size < 150 micron</p>
- > Temperature = 500 °c, 550 °c, 600 °c
- Flow rate = 75 ml/min, 100 ml/min, 125 ml/min
- Bed height = fully filled and monolayer

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••• Boat Reactor Geometry for Modeling



••• Governing Equations

$$\frac{\partial \mathbf{u}}{\partial t} + \rho(\mathbf{u} \cdot \nabla)\mathbf{u} = \nabla \cdot [-p\mathbf{I} + \tau] + \mathbf{F}$$
Navier Stokes equation
$$\frac{\partial c}{\partial t} + \mathbf{u} \cdot \nabla c = \nabla \cdot (D\nabla c)$$
Mass transport equation in quartz tube
$$\frac{\partial c}{\partial t} = \nabla \cdot (D\nabla c) + R$$
Mass transport equation in boat

Reaction: $C + O_2 \rightarrow CO_2$

••• Boundary Conditions



boundary no.	velocity	O ₂	CO ₂	char
1	u_in	c_o2_fluxin		NA
2	outlet	convecti	ve flow	NA
3,4,5		Zei	ro flux (wall	.)
6	wall	contir	nuity	no flux
all other		zero flu>	k (wall)	NA



domain no.	velocity	O ₂	CO ₂	char
1	0	Zero conc	centration	NA
2	NA	Zero conc	centration	c_cinit

Results:

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Time= 0 Surface: Concentration (molim*3) Time=500 Surface: Concentration (mol/m *3) A (%) ())) Fime=1003 Surface: Concentration (mol/m⁻³) 70 Time=1500 Surface: Concentration (maimi*3) M 50 Time=2500 Surface Concentration (motion ~3) 40 Time=3500 Surface: Concentration (mn//m*3) Time=4500 Surface. Concentration (molim*3) 100 Time=5500 Surface: Concentration (mol/m/13)

Evolution of char concentration in boat over time

Results: Outlet Gas Concentration



Qualitatively matches with experimental results !!

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••• Conclusion

- Effect of different operating parameters and bed diffusivity/bed height on the char combustion in UCG like condition are evaluated.
- Multiphysics modeling using COMSOL provided an insight of the experiments.
- Dependence on partial pressure of oxygen is to be determined and modeled.
- Modeling strategy is to be extended to real UCG conditions for early cavity growth.

Thank you !

••• References

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• • Extra Slides

	Name	Expression
	rho	(100000*32/1000/8.314/873) [kg/m^3]
	eta	(70e-6/rho) [Pa/s*kg/m^3]
	u_in	.05[m/s]
	D1_02	5e-5[m^2/s]
	D2_02	1e-5[m^2/s]
	c_cinit	(1/12e-3) [mole/m^3]
	D1_co2	5e-5[m^2/s]
	D2_co2	1e-5[m^2/s]
	c_o2_fluxin	.1[mol/m^2/s]
tor	psi	3.85
ιοι	k0	2e9
	E	95700
	R_g	8.314
	Т	600 [K]

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Name	Expression	
R_c	-k1*c_o2	
R_02	-k1*c_o2	
R_co2	k1*c_o2	
k1	ka1*(1-psi*log(1-X))^.5*flc2hs(X, 1e-3)	
Х	max((c_cinit-c_c)/c_cinit,0)	
ka1	k0*exp(-E/R_g/T)	



The boat reactor set up consists of

a cylindrical quartz tube (length 0.8 m and diameter 0.05 m), and

a rectangular quartz boat is placed inside the quartz tube. Boat dimensions are: length 0.15m, breadth 0.035m and thickness 0.005m.

	Dry basis	
Proximate	Volatile matter	44.92%
Analysis	Fixed Carbon	46.61%
_	Ash	8.47%
Ultimate	Carbon	40.594%
Analysis	Hydrogen	5.672%

Experimental Results



Figure 2 Concentration of CO2 and O2 vs. time at 500 and 550°C respectively (fig. a, b, c, d), comparison of monolaver and complete filling at 500°C, 100 ml/min (fig. e, f)

Assumptions:

□ Incompressible laminar flow

- □ Inlet gas is pure oxygen
- □ Flow of gas in the channel,
- \Box Diffusion and combustion in the boat.
- \square 2-D geometry modelled.

The mesh quality and quantity is as following: minimum element quality: 0.6689, average element quality: 0.9909, triangular elements: 151779.

